

Tetrathiafulvalene with a fused pyrazine ring

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Key indicators

Single-crystal X-ray study

T = 296 K

Mean $\sigma(\text{C}-\text{C}) = 0.016 \text{ \AA}$

R factor = 0.079

wR factor = 0.198

Data-to-parameter ratio = 16.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, 2-(1,3-dithiol-2-ylidene)-2,3-dihydro-1,3-dithia-4,7-diazaindene (pyrazinotetrathiafulvalene), $\text{C}_8\text{H}_4\text{N}_2\text{S}_4$, forms a head-to-tail type of π - π -stacking centrosymmetric dimer with an interplanar distance of 3.59 (1) Å . The dimers form a two-dimensional column along the [101] direction.

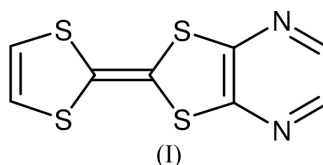
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Comment

It is known that $\text{S}\cdots\text{N}$ inter-heteroatom contacts work to construct unique molecular networks (Yamashita & Tomura, 1998). In the title compound, (I) (Fig. 1), short $\text{S}\cdots\text{N}$ inter-heteroatom contacts [3.10 (1) Å for $\text{S2}\cdots\text{N1}(x, y + 1, z)$] are observed between the two π - π dimers, as shown in Fig. 2. Two molecules in a dimer are related by an inversion center. A planar π -conjugated molecule prefers to overlap with a single molecule at both sides of the molecular plane to form a one-dimensional column. In the two-dimensional column of (I), however, one dimer bridges two other dimers through the overlap of the pyrazine and 1,3-dithiole rings (Fig. 3). The angle between the dimers is 111 (1) $^\circ$. This type of unique multi-dimensional structure is important for the construction of organic conducting materials (Barclay *et al.*, 2000; Kato *et al.*, 1988; Morimoto & Inabe, 1995; Tomura & Yamashita, 2000; Yamashita *et al.*, 1997, 1998).



Experimental

The title compound (I) was synthesized according to the literature method of Papavassiliou *et al.* (1988). Crystals of (I) were grown from a dichloromethane solution.

Crystal data

 $\text{C}_8\text{H}_4\text{N}_2\text{S}_4$ $M_r = 256.37$ Monoclinic, $P2_1/n$ $a = 8.481$ (2) Å $b = 9.1928$ (9) Å $c = 12.8980$ (14) Å $\beta = 96.89$ (2) $^\circ$ $V = 998.4$ (3) Å^3 $Z = 4$ $D_x = 1.706 \text{ Mg m}^{-3}$ Cu $K\alpha$ radiation

Cell parameters from 24 reflections

 $\theta = 9.7\text{--}42.8^\circ$ $\mu = 8.39 \text{ mm}^{-1}$

T = 296 (2) K

Needle, orange

 $0.40 \times 0.05 \times 0.05 \text{ mm}$

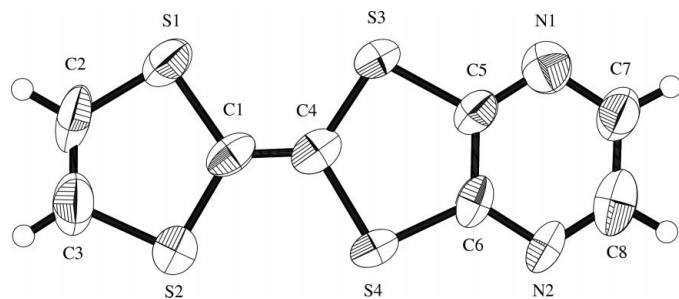


Figure 1
The molecular structure of (I) with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

Data collection

Enraf–Nonius CAD-4 diffractometer
 ω scans
 Absorption correction: ψ scan (North *et al.*, 1968)
 $T_{\min} = 0.134$, $T_{\max} = 0.679$
 2060 measured reflections
 2026 independent reflections
 663 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.084$
 $\theta_{\text{max}} = 74.2^\circ$
 $h = -10 \rightarrow 10$
 $k = 0 \rightarrow 11$
 $l = 0 \rightarrow 16$
 3 standard reflections
 frequency: 120 min
 intensity decay: 4.9%

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.079$
 $wR(F^2) = 0.198$
 $S = 1.07$
 2026 reflections
 127 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0473P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.47 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.62 \text{ e } \text{\AA}^{-3}$

All H atoms were placed in geometrically calculated positions and refined by using a riding model with C–H distances set to 0.93 Å.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1992); cell refinement: *CAD-4 EXPRESS*; data reduction: *HELENA* (Spek, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL97*.

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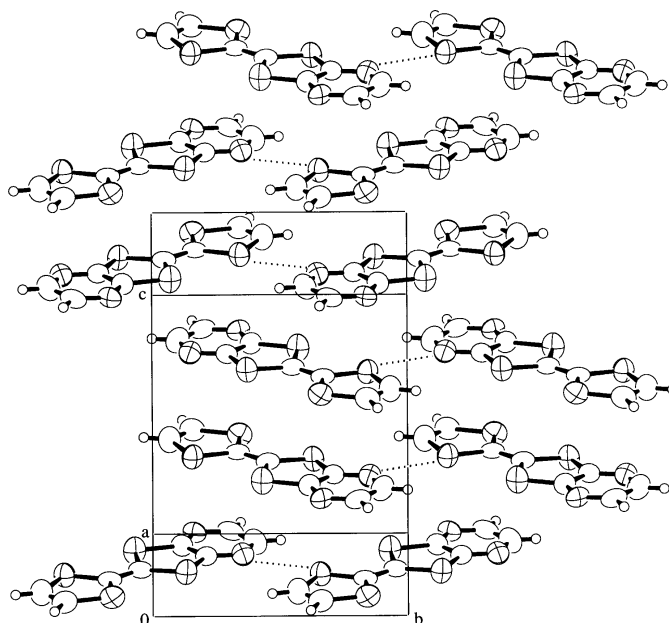


Figure 2
Packing diagram of (I). Short S...N inter-heteroatom contacts are indicated by dotted lines.

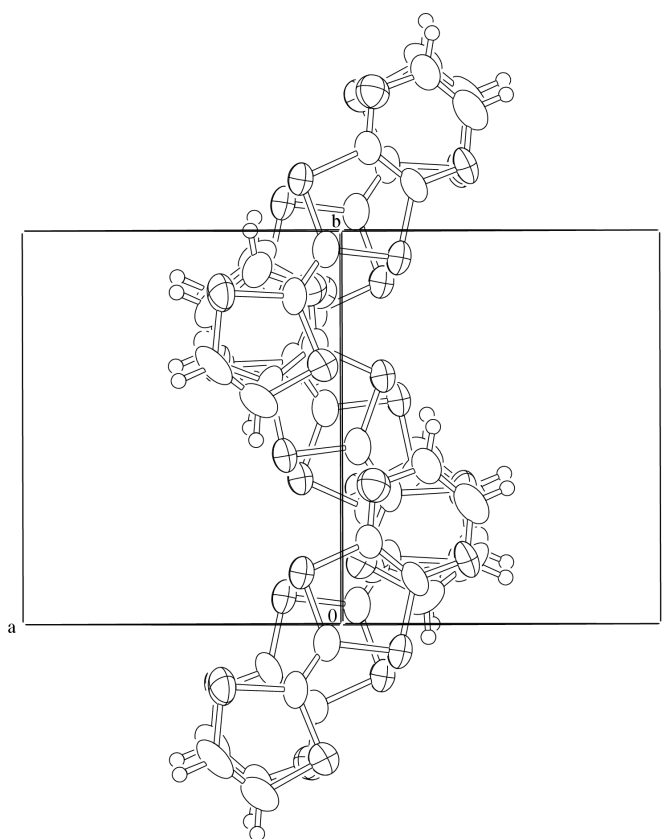


Figure 3
Packing diagram of (I) viewed along the [101] direction. Three dimers (six molecules) are indicated.

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